

# Chapter E1: Basic Electrostatics

## E1.3 What is the Nature of Charge?

- Charge can be “positive” or “negative”.
- Like charges repel. Opposites attract.

## E1.4 How Objects Become Charged

The SI unit of charge:

$$1\text{C} \equiv \text{charge of } 6.242 \times 10^{18} \text{ protons} = (6.242 \times 10^{18})e \quad (1)$$

where

$$e \equiv \text{charge of one electron} = 1.602 \times 10^{-19}\text{C} . \quad (2)$$

## E1.5 Conservation of Charge

Charge is conserved.

## E1.6 Coulomb’s Law

Coulomb’s law says:

$$F_e = \frac{k|q_1q_2|}{r^2} . \quad (3)$$

What do the symbols in this equation stand for?

Under what circumstances does this equation apply?

## **E1.7 Conductors and Insulators**

The difference between insulators and conductors:

- insulators
- conductors

## **E1.8 Polarization of Neutral Objects**

**Practice:**

Four identical particles, each having a charge of  $+120 \text{ nC}$  are arranged in a square  $6.0 \text{ cm}$  on each side. Find the magnitude of the force acting on any one of the charges, and describe the direction of this force in words.